



# Welcome and Background

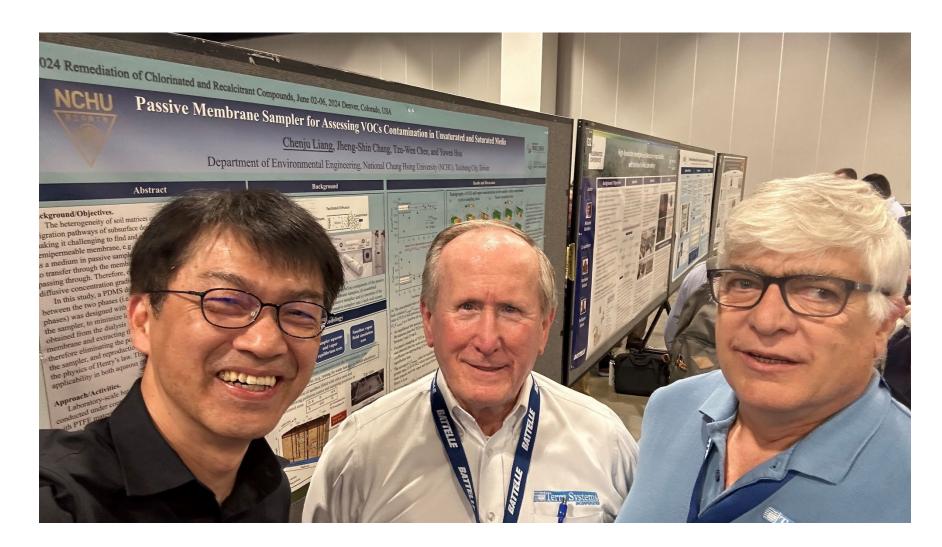


- Who Am I?
  - Career start in 1984 in original LNAPL recovery firm
  - Multiple firms as industry consolidated and changed
  - Focus on remediation, In-Situ Chemical Oxidation and Safety
  - Global technical presenter for last 40 years
- Why am I here?
  - Supporting local projects in Taiwan ISB and ISCO
  - Met Dr. Liang at Battelle 2024 while reviewing his poster and research
  - Information sharing!





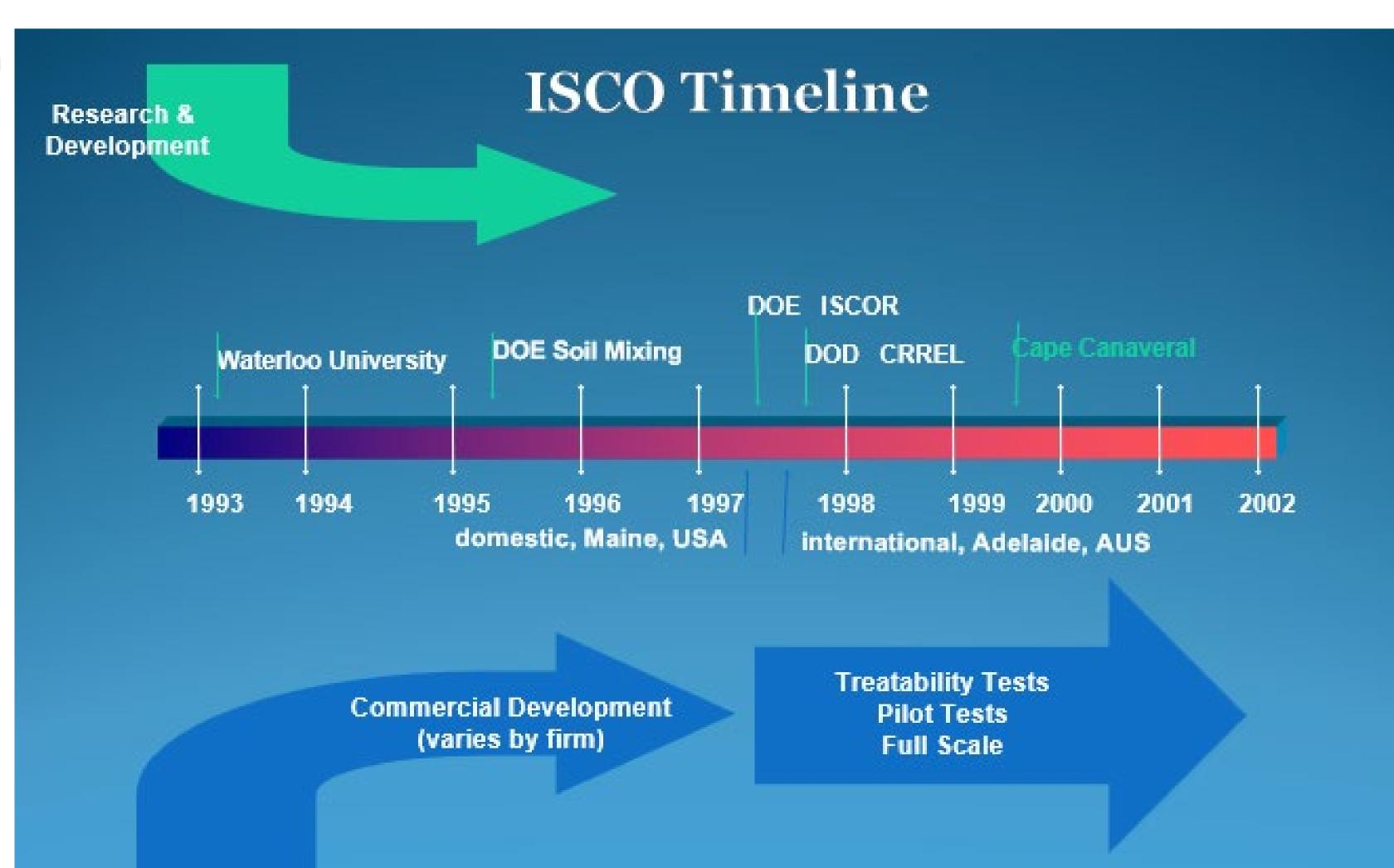




### What is In-Situ Chemical Oxidation (ISCO)



- ISCO is another remedial tool in the remedial toolbox
  - Highly flexible and modifiable
  - Not applicable to any Site (no silver bullet)
- Accepted mainstream technology (no longer considered innovative, some even consider "traditional")
  - Educational ISCO ~ 1994
  - DOE Soil Mixing ~1996
  - First Field ISCO 1997
  - Commercialization post 1997



## What is In-Situ Chemical Oxidation (ISCO)



- ISCO is the application of chemical oxidants to destroy contaminants in the subsurface.
- The oxidant is the chemical applied to mineralize (i.e., destroy) target compounds into innocuous byproducts.
- Oxidation is the chemical reaction where electrons are lost by the reduced species and absorbed by the oxidant (i.e., redox reactions):

$$MnO_4^- + 4H^+ + 3e^- \rightarrow MnO_2 + 2H_2O$$

 Applying this half reaction to various contaminants, the corresponding "stoichiometric" requirements for oxidation can be calculated.

# In-Situ Chemical Oxidation (ISCO)





2. Delivery and Distribution

Performance is:

applying enough oxidant in contact with the contaminant for

a long enough period of time to react effectively

3. Contact and Persistence

4. Reaction and Contaminant Destruction

# Reactivity of Available Oxidants

3.0 -

2.75

2.5

2.25

2.0

1.75

1.5

1.25

1.0

Hydroxyl Radical

Persulfate Radical

Ozone Persulfate

Hydrogen Peroxide Permanganate Bleach

Chlorine gas

Oxygen

Oxidation of Organics

C-C (Sat'd)

PCBs 🔘



Oxidation of Inorganic

#### Oxidant Selection

#### Selection is a function of many factors:

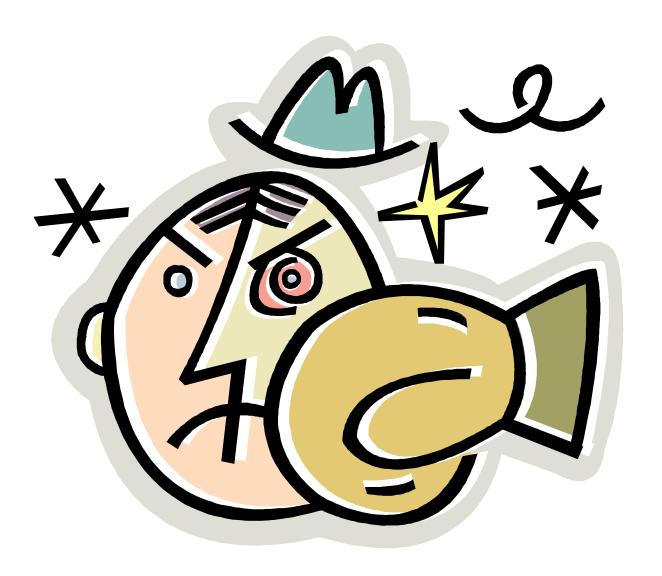
- Site specific issues lithology, groundwater hydraulics, access, sensitive receptors, target mass location, heterogeneity and anisotropy
- Time available, single vs. multiple treatments, permeability
- Cost labor, equipment, capital, expenses, subcontractors, availability
- Oxidant loading, effectiveness, stability, end products of reaction
- Optimization strategies surfactants, coupling, controlled release
- Focus is on permanganate ISCO, technology is however applicable to many injectates

# Delivery



# Delivery is, remains, and always will be, the single most important component of an ISCO success or failure

- Contact contact contact! redox is contact process
- Emplacement is only the beginning –
- End point persistence vs. reactivity





# Increasing Disruption

# Delivery Examples



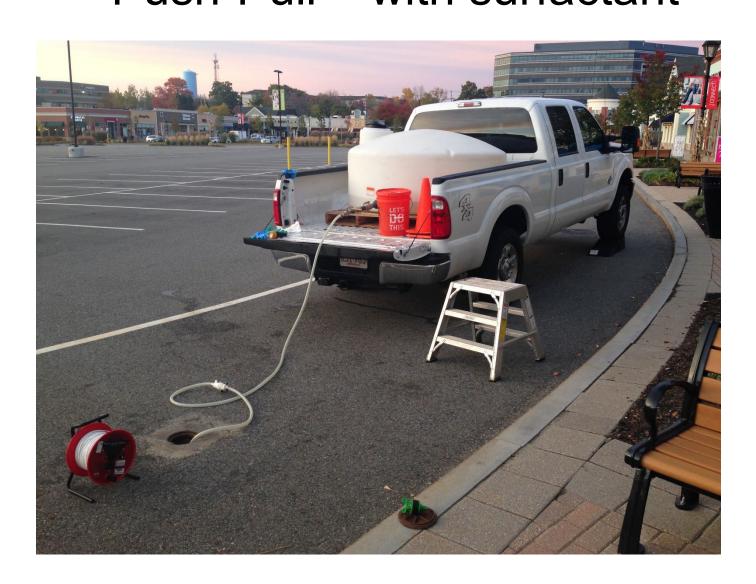
"Passive"	"Active"	"Destructive"
Encapsulation	Existing Wells Temporary Wells	Excavation / Engineering
Flow through	Recirculation Pore dilation	Media Fracturing
Constant head	Electrokinetics	Soil Mixing
Push-Pull	Thermal Methods	

#### Passive Methods

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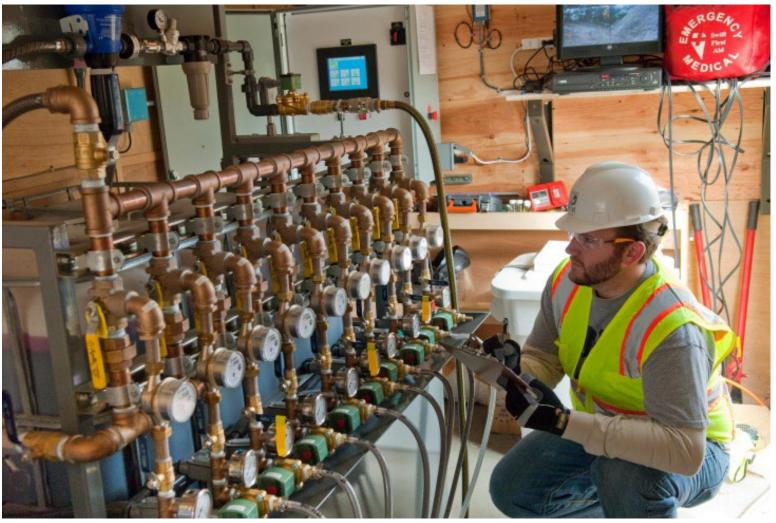
#### Allow unimpeded natural processes

- Do nothing or very little
- Encapsulation/isolation Cap and cover
- Flow through systems PRBs, mulch walls
- Dissolvables candles
- Contact Head injection –
- Push-Pull with surfactant









#### Active Methods



#### Accelerate Natural Processes

- Pressurized injection wells, probes, multiple level
- Recirculation P&T and infiltrate, ART wells
- Pore Dilation –
- Electro kinetics -
- Thermal (?) -







#### Destructive Methods

Site conditions get in the way – destroy them

- Excavation
- Soil Mixing
- Media Fracturing







# Oxidant Misconceptions



- In porous media, the application of oxidants **will not "clog"** the formation (< 10% void infilling at high loading). Due to small pore throats and limited connectivity, this is not necessarily true in low permeability materials or fractured media
- Oxidation does not sterilize formation, biota recovers quickly and fully
- Effective and appropriate delivery is critical to ISCO success
- Oxidant can persist in presence of contaminant
- Excess oxidizer lead to false negative confirmatory samples (also may damage to lab equipment)
- Treatment trains biological methods (ERD using QRS™ or SRS®) provide follow-up to ISCO

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#### ISCO with Sodium Persulfate



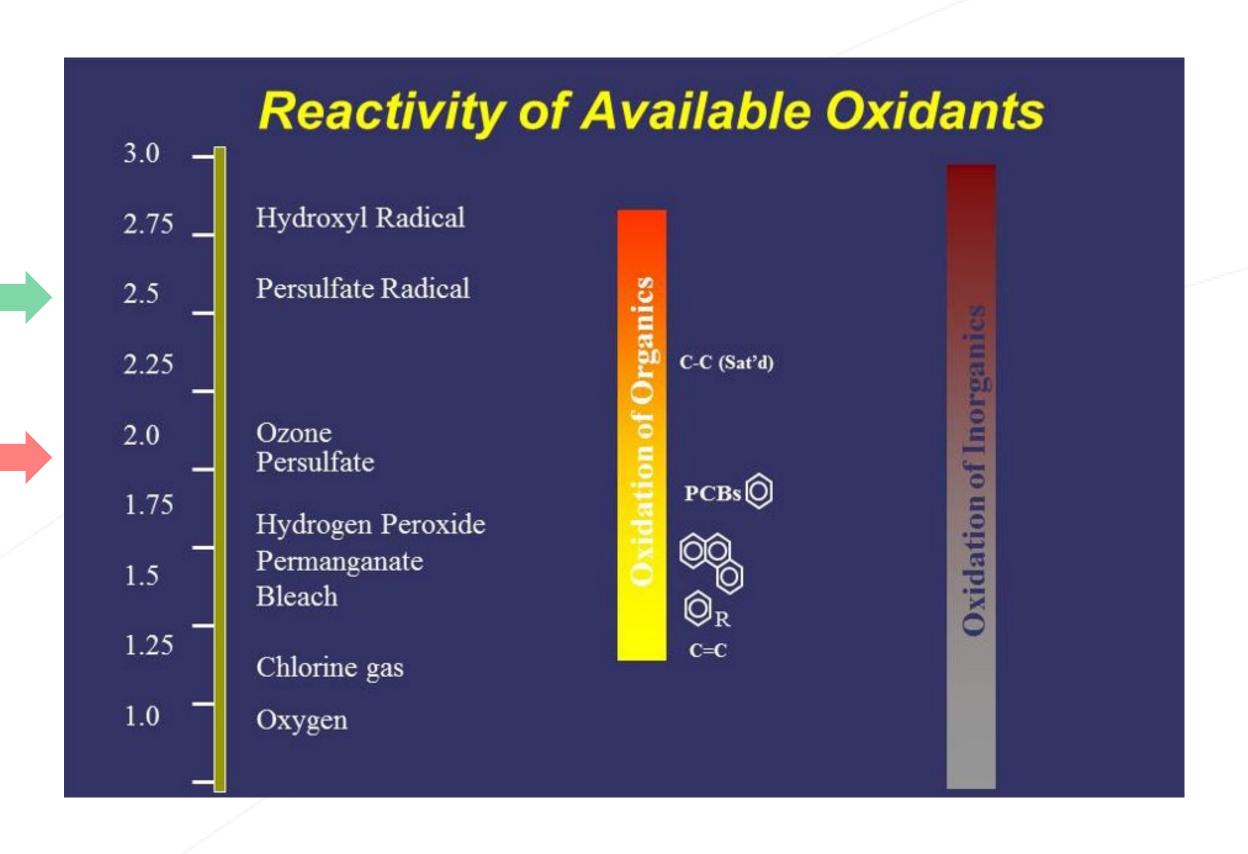
Two Reaction Pathways

activated (radical) - 
$$S_2O_8^{-2}$$
 + TSI-FSA<sup>TM</sup>  $\rightarrow S_2O_8^{-}$ ,  $SO_4^{-\bullet}$ , OH•,  $O_2^{-\bullet}$ 

Why Activation?

unactivated (oxidation) – 
$$S_2O_8^{-2}$$
 + 2H<sup>+</sup> + 2e<sup>-</sup>  $\rightarrow$  2H<sub>2</sub>SO<sub>4</sub><sup>-</sup>

- 1. combined oxidative and reductive pathways
- 2. stronger, kinetically fast, localized
- 3. degrade recalcitrant compounds (Cl-ethanes, CT, pesticides, etc.)



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# Commonly Used Persulfate Activation Methods



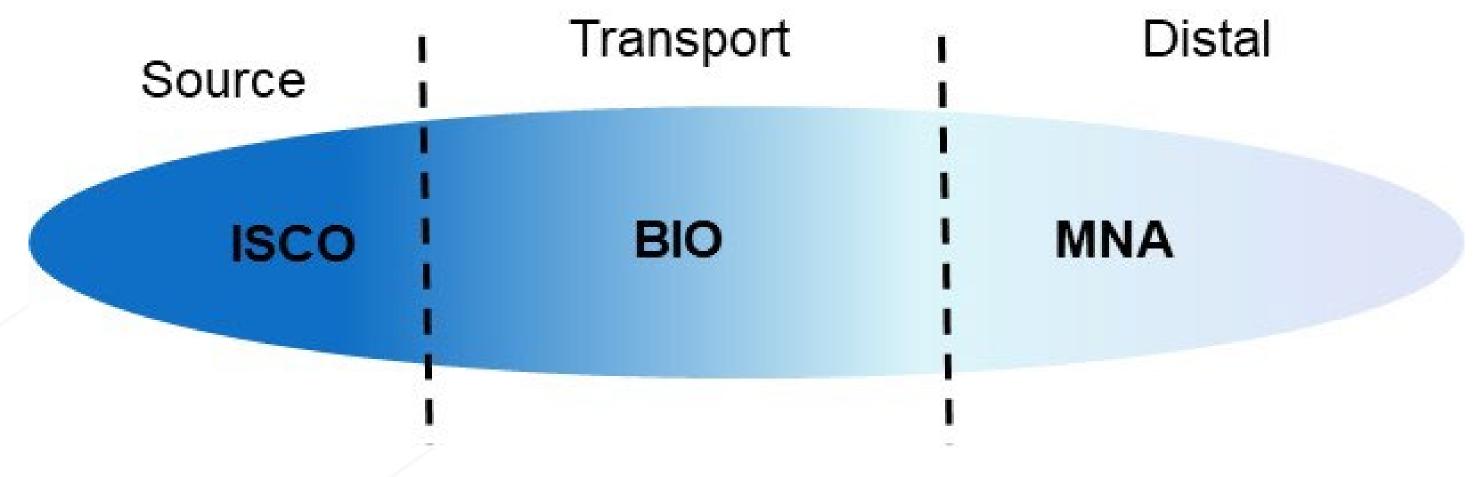
Method	Drawbacks	Patent Holder
Sodium Hydroxide	Hazardous (pH, worker safety)	Evonik (Block et al.)
Fe-EDTA	Requires 160 to 600 mg/L Iron	Evonik (Block et al.)
Hydrogen Peroxide	Rapid decomposition of persulfate and peroxide	Evonik (Sethi et al.)
Stabilized Hydrogen Peroxide	Precise monitoring and control of pressure and temperature	Cronk (JAG)
Carbohydrate	Licensed technology	ISOTEC (Watts et al.,)
Ferrous Sulfide TSI-FSA <sup>TM</sup>	Limited field experience	None

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# Combined Remedies: Chemical Oxidation & Bioremediation



- Technology Coupling "Best of Both Worlds"
  - maximize effectiveness while minimizing cost
  - spatially sequentially over distance
  - temporally sequentially over time
  - better respond to changes in conditions
  - mutually supportive
- Remedial Life Cycle Approach



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#### State of the Practice

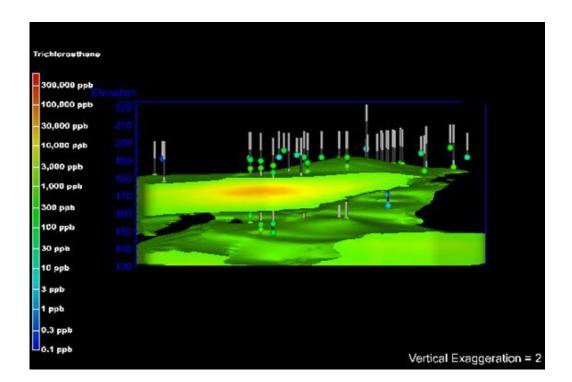
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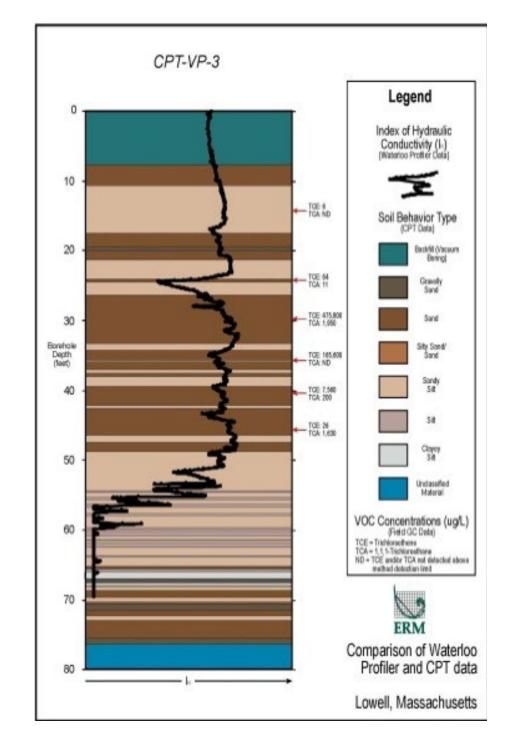
- Sites are increasingly complex
- Rapidly changing characterization tools
- Improved ISCO understanding
- Overpromised unrealistic expectations
  - Multiple incidences of technology failures
  - No technology works everywhere

#### Something you never want to see...











#### State of the Practice

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"... because as we know, there are **known knowns**; there are things we know we know.

We also know there are **known unknowns**; that is to say we know there are some things we do not know.

But there are also **unknown unknowns** – the ones we don't know we don't know.

And if one looks throughout the history of our country and other free countries, it is the latter category that tend to be the difficult ones,"

(Rumsfeld, 2002)



# Questions?





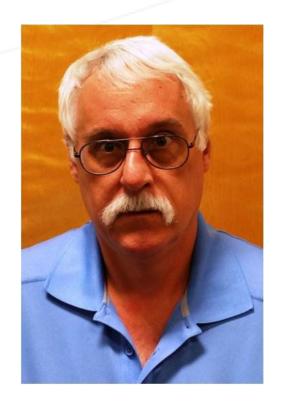
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